



Molecular Crystals and Liquid Crystals Incorporating Nonlinear Optics

Publication details, including instructions for authors and
subscription information:

<http://www.tandfonline.com/loi/gmcl17>

Synthesis and Properties of Ferroelectric 4-[4-(S-1- Methylheptyloxy)Benzoyloxy]-4'-n- Alkyloxycarbonylbiphenyls

O. Adomėnienė^a, P. Adomėnas^a, R. Bernotas^a, J. Petraitis^a &
M. Jakubėnienė^a

^a Vilnius University, Lithuania

Version of record first published: 22 Sep 2006.

To cite this article: O. Adomėnienė, P. Adomėnas, R. Bernotas, J. Petraitis & M. Jakubėnienė (1990): Synthesis and Properties of Ferroelectric 4-[4-(S-1-Methylheptyloxy)Benzoyloxy]-4'-n-Alkyloxycarbonylbiphenyls, *Molecular Crystals and Liquid Crystals Incorporating Nonlinear Optics*, 191:1, 187-191

To link to this article: <http://dx.doi.org/10.1080/00268949008038592>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.tandfonline.com/page/terms-and-conditions>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

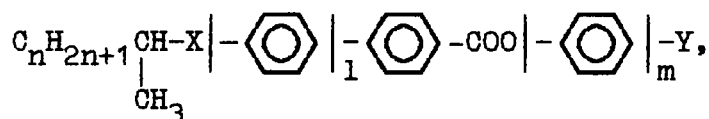
SYNTHESIS AND PROPERTIES OF FERROELECTRIC 4-[4-(S-1-METHYLHEPTYLOXY)BENZOYLOXY]-4'-n- ALKYLOXYCARONYLBIPHENYLS

O.ADOMENIENE, P.ADOMENAS, R.BERNOTAS,
 J.PETRAITIS and M.JAKUBENIENE
 Vilnius University, Lithuania

Abstract Synthesis, mesomorphic properties
 and spontaneous polarization values of
 4-[4-(S-1-methyl-heptyloxy)benzoyloxy]-
 -4'-alkyloxy-caronylbiphenyls are given.

INTRODUCTION

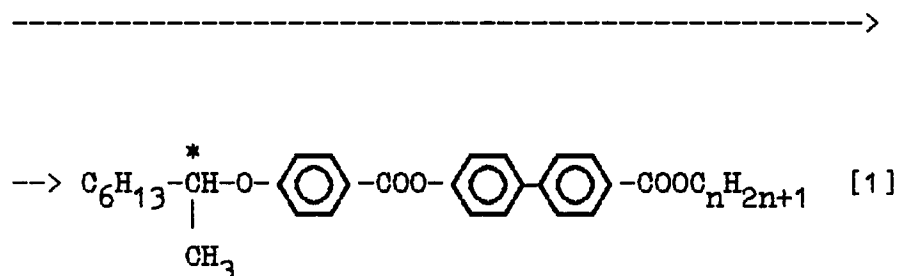
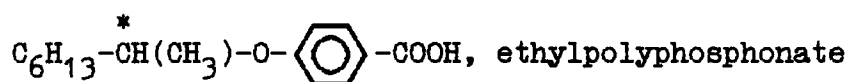
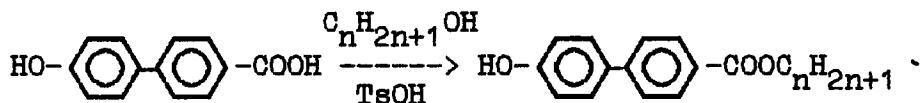
The patent¹ of Chisso company is known, which
 claims the compounds



where $n=2\div 16$, $l=0, 1$ or 2 , $X=-\text{CH}_2-$ or $-\text{O}-$, $\text{Y}=-\text{R}$,
 $-\text{OR}$, $-\text{COR}$, $-\text{OCOR}$, $-\text{OCOOR}$ or $-\text{COOR}$. The authors
 indicate the high stability and small response
 times, but no examples of numerous classes of
 claimed compounds are presented. We have
 synthesized the homologous 4-[4-(S-1-methyl-
 heptyloxy) benzoyloxy]-4'-n-alkyloxy- caronyl-
 biphenyls. The phase transition temperatures, the
 values of spontaneous polarization and the response
 times of electrooptical cells are presented.

RESULTS

The 4-[4-(S-1-methylheptyloxy)benzoyloxy] 4'-n-alkyloxycarbonylbiphenyls were synthesized according following scheme:



The compounds [1] exhibit the transition from crystal form to ferroelectric smectic C phase at moderate temperatures. Most of the compounds [1] form different crystal modifications when crystallized from different solvents (higher melting points are obtained from ethanol, the lower ones - from hexane). The phase transition temperatures are given in Table 1.

The values of spontaneous polarization have been determined by Sawyer-Tower method in the cells with an interelectrode gap 10 μm , they are given in Table 1. These values correspond to the temperatures, by 10 $^{\circ}\text{C}$ lower than the phase transition points (S_C-S_A or S_C-N). The tilt angle is $\theta=30^{\circ}$ ($n=6$, 46°C), 32° ($n=8$, 60°C).

The same cells have been used both for spontaneous polarization and electrooptical measurements. The cells have been placed between the crossed polarizers, the electrodes being connected with the source of the rectangular voltage pulses with an alternation of polarity at 1 Hz frequency. The response less than 1 ms is observed when voltage is 10V (see Table 1).

TABLE 1 The phase transition temperatures ($^{\circ}\text{C}$), the values of spontaneous polarization ($P_s, \text{nC/cm}^2$) and electrooptical cell response times (τ, ms) of 4-[4-(S-1-methylheptyloxy)benzoyloxy]-4'-n-alkyloxybiphenyls.

n	Cryst solvent	K	S_C	S_A	N	$I P_s$	τ_{on}	τ_{off}
5	Hexane	.40.5	.	-	66	.78	.58	9.7 1,5
6	Hexane	.45	.	-	63	.72.5.	.84	0.8 0.5
	EtOH	.52	.	-	63	.72.5.		
7	EtOH	.53	.	-	64	.76	.77	1.0 0.5
8	EtOH	.48,5	. 66	.	67	.74.5.	.65	0.8 0.3
9	Hexane	.46	. 64	.	67	.76.5.	.58	0.9 0.6
	EtOH	.50	. 64	.	67	.76,5.		
10	Hexane	.38	. 63	.	66	.73	.	
	EtOH	.53	. 63	.	66	.73	.	
11	Hexane	.52	. 64	.	69	.72	.	
12	Hexane	.49	. 62	.	64	.69	. 51	1,1 0.9

EXPERIMENTAL

4-Alkyloxycarbonyl-4'-hydroxybiphenyls

A mixture of 0.02 mol (4.3 g) 4-hydroxybiphenyl-4'-carboxylic acid, 0.1 mol n-alkanol, 150 ml toluene and 0.5g p-toluenesulfonic acid was refluxed with azeotropic elimination of water until dissolving all the 4-hydroxy biphenyl-4'-carboxylic acid (6-10 h). Toluene (200 ml) was added, the solution was washed with water and dried. The solvent and the excess of n-alkanol were evaporated at a reduced pressure, the residue was recrystallized from mixture of toluene and hexane (4:1). The yield (%), melting point (°C) and n are given: 56, 105-106, 5; 67, 99-101, 6; 53, 97-98, 7; 55, 101-103, 8; 60, 98-100, 9; 54, 109-110, 10; 53, 101-103, 11; 52, 104-105, 12.

The structure of the compounds was confirmed by PMR spectra and elementary analysis data.

4-[4-(1-S-Methylheptyloxy)benzoyloxy]-4'-n-alkyloxycarbonylbiphenyls

A mixture of 0.01 mol 4-alkyloxycarbonyl-4'-hydroxybiphenyl, 0.01 mol (2.50 g) 4-(S-1-methylheptyloxy)-benzoic acid, was refluxed for 30h. The solvent was evaporated, the residue was diluted by 50 ml water and extracted with diethyl ether. The extract was washed by water, dried and the solvent was evaporated. The residue was chromatographed on silicagel, an eluant - a mixture of hexane with diethyl ether (1:1). The solvent was evaporated, the residue was crystallized, by turns, from hexane and ethanol

till constant phase transition temperatures. The structure of all compounds [1] was confirmed by PMR spectra and elementary analysis data.

REFERENCES

1. H.Inoue, S.Saito, K.Miyazawa, T.Inukai and K.Terashima, Eur. Pat. Appl. 0164814 A2(1985).